

## Chemical Formula and Equations - Common Chemical Formula List

The following list of Chemical Formula and Equations shows some of the most common chemical formulas.

Common Chemical Formula List	
Na	Sodium
H <sub>2</sub> O	Water
C <sub>6</sub> H <sub>12</sub> O <sub>6</sub>	Glucose
C <sub>2</sub> H <sub>6</sub> O	Alcohol
CaSO <sub>4</sub>	Sulphate Group
H <sub>2</sub> S	Hydrogen Sulphide
NaCl	Salt
O <sub>2</sub>	Oxygen
C <sub>2</sub> H <sub>6</sub> O	Ethanol
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Vinegar
NH <sub>3</sub>	Ammonia
Mg	Magnesium
C <sub>2</sub> H <sub>4</sub> O <sub>2</sub>	Acetic Acid
C <sub>4</sub> H <sub>10</sub>	Butane
NO <sub>3</sub> <sup>-</sup>	Nitrate
Cu	Copper
N <sub>2</sub>	Nitrogen
CO <sub>2</sub>	Carbon Dioxide
H <sub>2</sub> SO <sub>4</sub>	Sulphuric Acid

$\text{CH}_4$	Methane
$\text{C}_{12}\text{H}_{22}\text{O}_{11}$	Sucrose
$\text{C}_3\text{H}_8$	Propane
$\text{NaHCO}_3$	Baking Soda
F	Fluoride
$\text{F}_2$	Fluoride
$\text{H}_2\text{O}_2$	Peroxide
$\text{C}_8\text{H}_{10}\text{N}_4\text{O}_2$	Caffeine
$\text{NaCl}$	Sodium Chloride
$\text{C}_9\text{H}_8\text{O}_4$	Aspirin
HCl	Hydrochloric Acid
$\text{Zn}(\text{NO}_3)_2$	Zinc
CO	Carbon Monoxide
$\text{NaOH}$	Sodium Hydroxide
$\text{NaCN}$	Sodium Cyanide
$\text{Ca}(\text{CN})_2$	Calcium Cyanide
Au	Gold
I	Iodine
Sn	Tin
$\text{C}_6\text{H}_6$	Benzene
$\text{H}_2\text{S}$	Hydrogen Sulfide
$\text{CH}_3\text{COCH}_3$	Acetone
$\text{H}_3\text{PO}_4$	Phosphoric Acid
$\text{C}_5\text{H}_{12}$	Pentane
$\text{CH}_3\text{OH}$	Methanol

<b>HBr</b>	<b>Hydrobromic Acid</b>
<b>H<sub>2</sub>CO<sub>3</sub></b>	<b>Carbonic Acid</b>
<b>Ti</b>	<b>Titanium</b>
<b>NaClO</b>	<b>Sodium Hypochlorite</b>
<b>C<sub>2</sub>H<sub>6</sub></b>	<b>Ethane</b>
<b>(NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub></b>	<b>Ammonium Sulfate</b>
<b>C<sub>8</sub>H<sub>18</sub></b>	<b>Octane</b>
<b>CuSO<sub>4</sub></b>	<b>Copper Sulfate</b>
<b>C<sub>27</sub>H<sub>46</sub>O</b>	<b>Cholesterol</b>
<b>C<sub>7</sub>H<sub>6</sub>O<sub>2</sub></b>	<b>Benzoic Acid</b>
<b>H<sub>2</sub>SO<sub>3</sub></b>	<b>Sulfurous Acid</b>
<b>C<sub>6</sub>H<sub>12</sub>O<sub>6</sub></b>	<b>Galactose</b>
<b>C<sub>6</sub>H<sub>8</sub>O<sub>6</sub></b>	<b>Ascorbic Acid</b>
<b>CO<sub>2</sub></b>	<b>Dry Ice</b>
<b>NaNO<sub>3</sub></b>	<b>Sodium Nitrate</b>
<b>CaO</b>	<b>Calcium Oxide</b>
<b>HIO<sub>3</sub></b>	<b>Iodic Acid</b>
<b>C<sub>3</sub>H<sub>6</sub>O<sub>3</sub></b>	<b>Lactic Acid</b>
<b>MgBr<sub>2</sub></b>	<b>Lactic Acid</b>
<b>H<sub>2</sub>O</b>	<b>Water Vapor</b>
<b>No Formula</b>	<b>Oxide</b>
<b>C</b>	<b>Carbon</b>
<b>H</b>	<b>Hydrogen</b>
<b>C<sub>n</sub>H<sub>2n</sub>O<sub>n</sub></b>	<b>Sugar</b>
<b>O<sub>4</sub>S<sup>2-</sup></b>	<b>Sulfate</b>

$\text{N}_2\text{O}$	Nitrogen
$\text{C}_6\text{H}_8\text{O}_7$	Citric Acid
$\text{C}_8\text{H}_{18}$	Octane
$\text{C}_{10}\text{H}_{16}\text{O}$	Camphor
$\text{AgI}$	Silver Oxide
$\text{As}_4\text{O}_3$	Arsenic Trioxide
$\text{Au}_2\text{O}_3$	Gold Trioxide
$\text{Au}_2\text{S}$	Gold Sulfide
$\text{Br}_2$	Bromine
$\text{Al}_2\text{O}_3$	Aluminium foil